**Polarity of molecules (Level 2) examiners tips: Read these please!**

• RTQ2 (read the question twice)

• this question is asking if the molecules are polar or non-polar

• draw a quick Lewis sketch – remembering the non-bonding/lone pairs of electrons

• think about the electronegativity difference between the atoms

• bear in mind the shape of the molecule

• then, look carefully to decide on your answer...polar molecules have an uneven distribution of charge

• view electron clouds of molecules in the online animations to help your visual understanding

**Also…”don’t be daft”**

rush your thoughts about your answer, take your time and think before you write down your answer

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